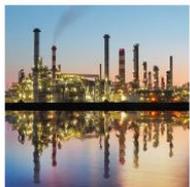


# Thermal desorption applications



# Application areas



Environmental monitoring

Biological profiling

Automotive studies

Defence & Forensics

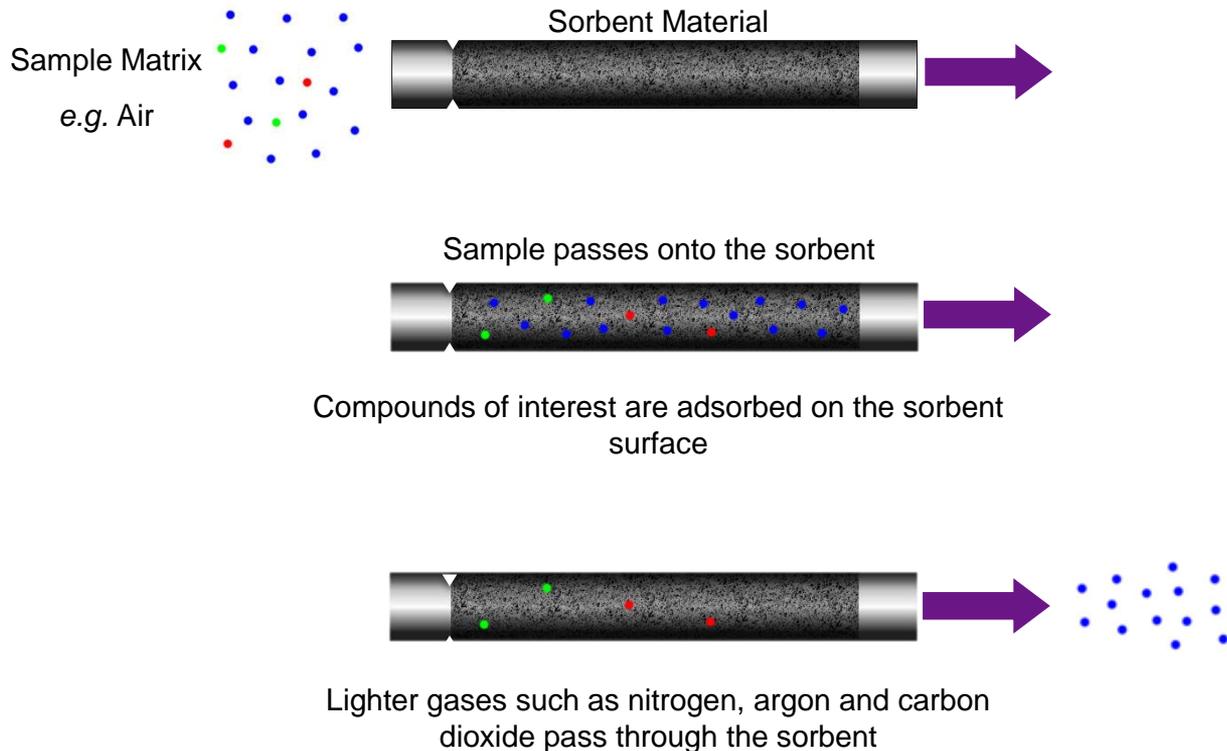
Fragrance & odour profiling

Consumer environmental health

Food & Drink

# What is thermal desorption (TD)?

Sample collection



# Challenges with air sampling

1. Many compounds are harmful at very low concentrations
  - ✓ Pre-concentration (Thermal Desorption)



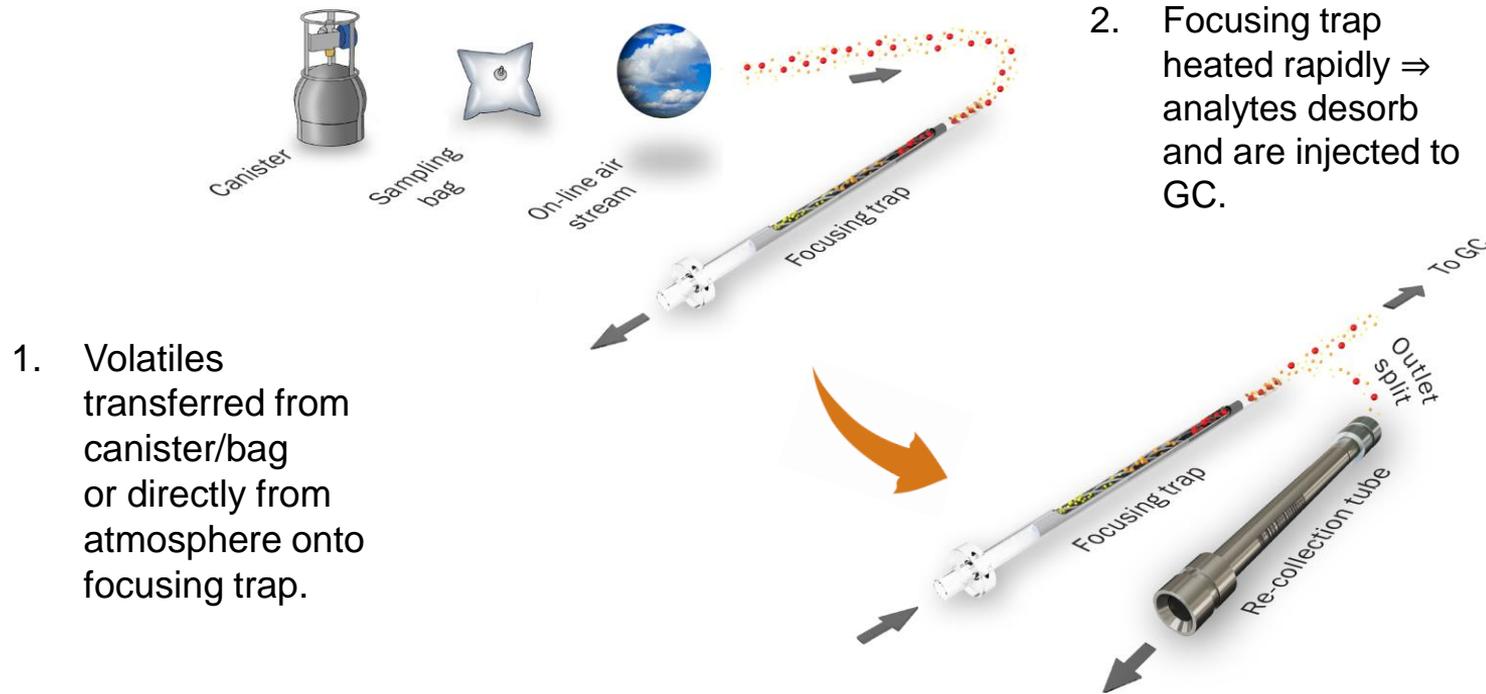
# Challenges with air sampling

1. Many compounds are harmful at very low concentrations
  - ✓ Pre-concentration (Thermal Desorption)
2. Many pollutants are very volatile – how can we collect and concentrate them?

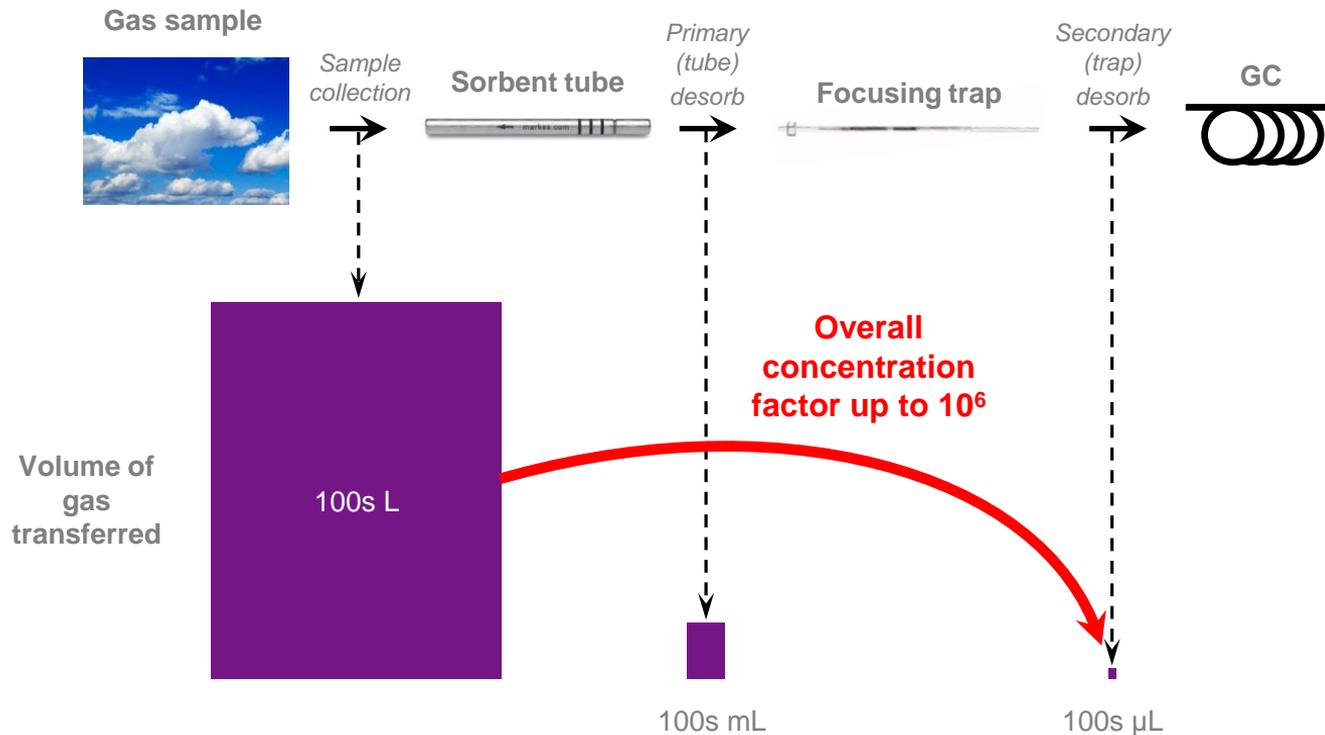


# 'Whole air' sampling for very volatile samples

How it works

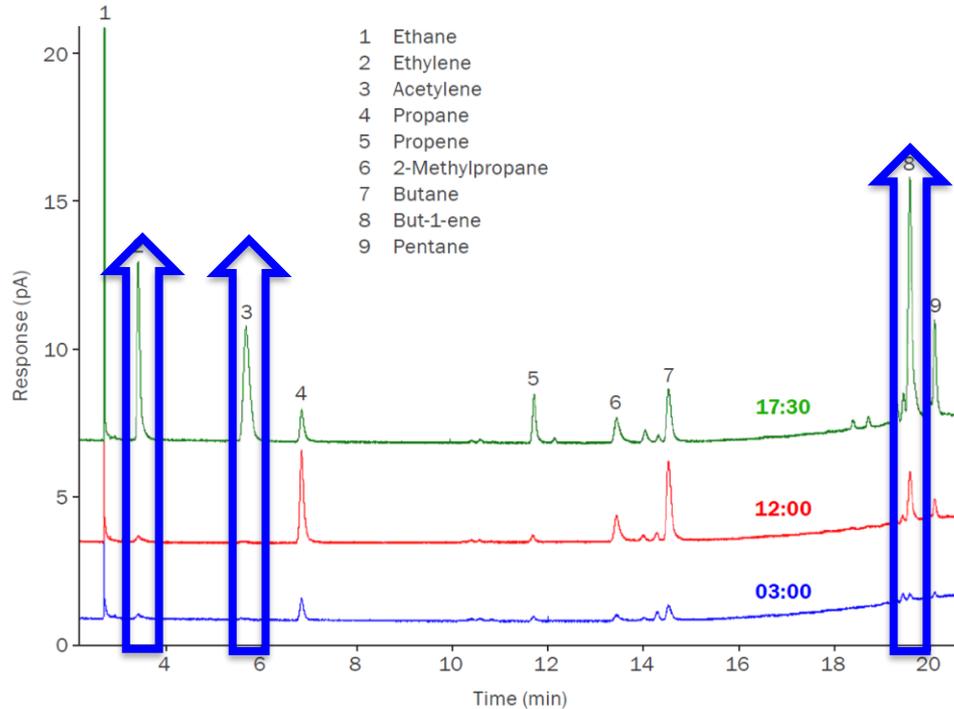


# What is thermal desorption?



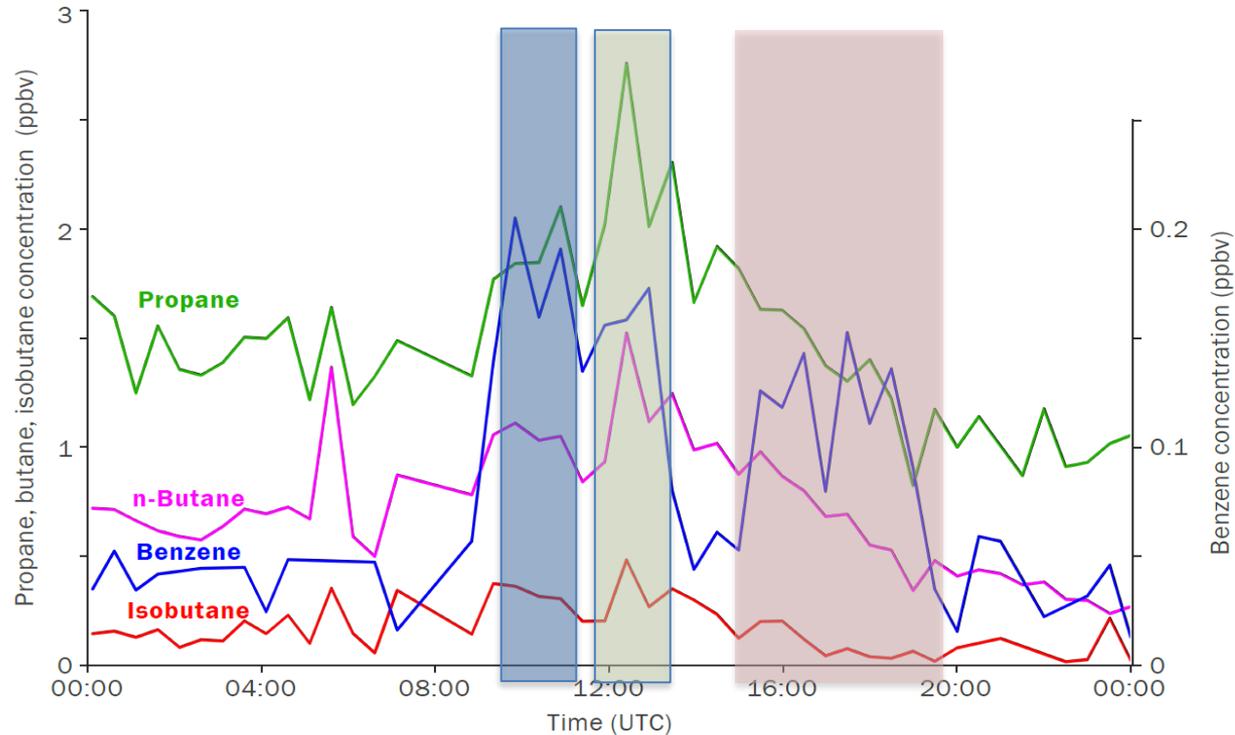
# On-line pollution monitoring

Very volatile ozone precursors



# On-line pollution monitoring

## Diurnal variation in pollutant concentrations



# Challenges with air sampling

1. Many compounds are harmful at very low concentrations
  - ✓ Pre-concentration
2. Many pollutants are very volatile – how can we collect and concentrate them?
  - ✓ Cryogen – free on-line or canister sampling
  - ✓ Selection of the optimum sorbent configurations



# Challenges with air sampling

1. Many compounds are harmful at very low concentrations
  - ✓ Pre-concentration
2. Many pollutants are very volatile – how can we collect and concentrate them?
  - ✓ Cryogen – free on-line or canister sampling
  - ✓ Selection of the optimum sorbent configurations
3. Air samples can be very humid – how do we remove water without removing compounds?

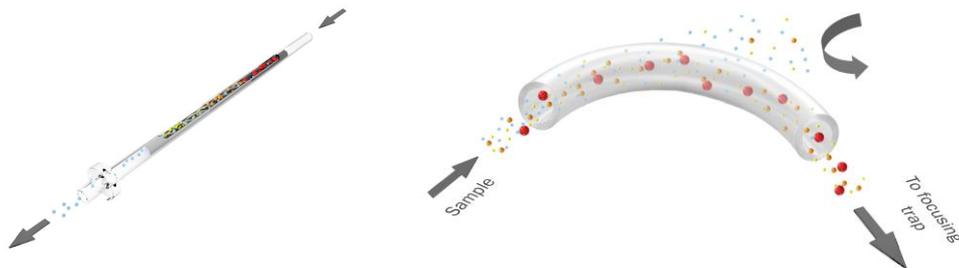


# Water management techniques for air samples



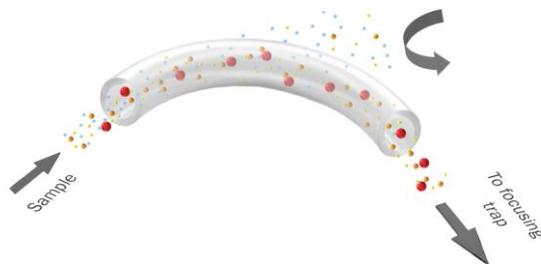
Analyte type	Dry purge only
C2 compounds (VVOC)	✗
C3+, Non-polar VOCs	✓
Pinenes	✓
Polar VOCs	✓
Sulphur compounds	✓

# Water management techniques for air samples



Analyte type	Dry purge only	Membrane dryer
C2 compounds (VOC)	✗	✓
C3+, Non-polar VOCs	✓	✓
Pinenes	✓	✗
Polar VOCs	✓	✗
Sulphur compounds	✓	✓

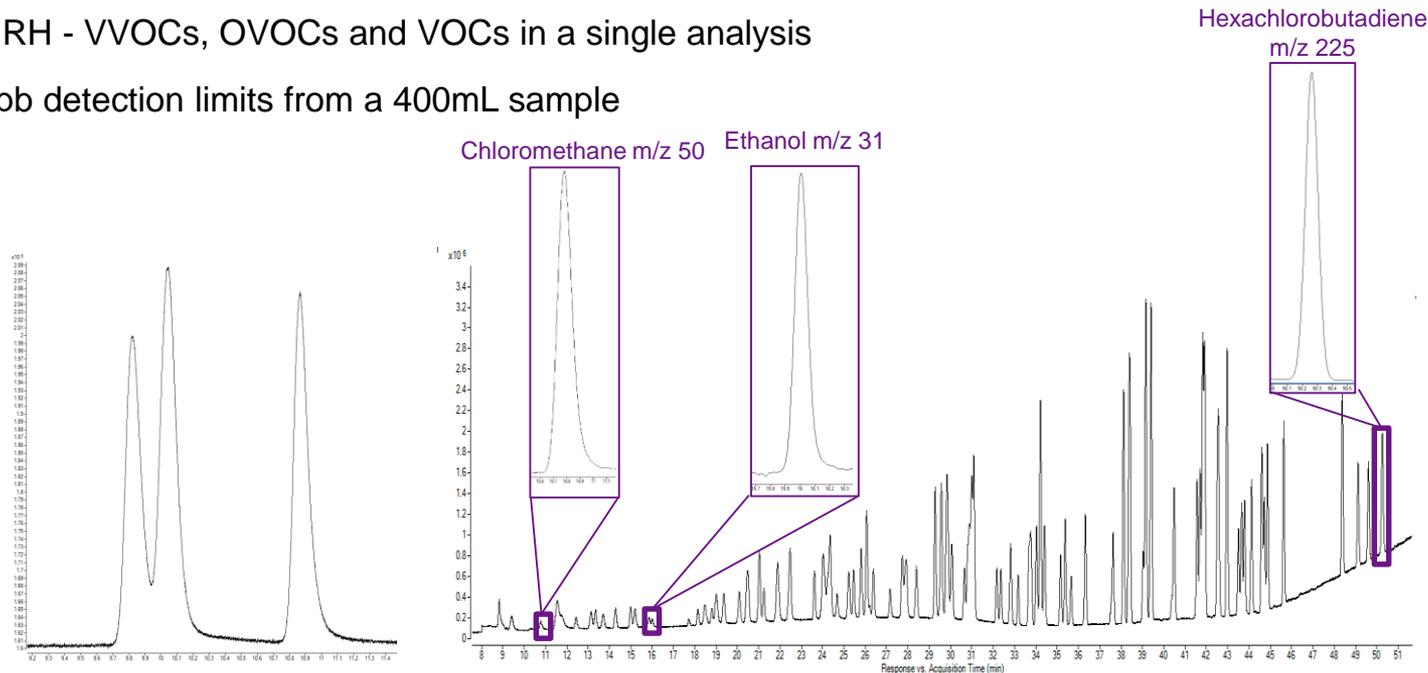
# Water management techniques for air samples



Analyte type	Dry purge only	Membrane dryer	Kori-xr
C2 compounds (VVOC)	✗	✓	✓
C3+, Non-polar VOCs	✓	✓	✓
Pinenes	✓	✗	✓
Polar VOCs	✓	✗	✓
Sulphur compounds	✓	✓	✓

# Cryogen-free combined ozone precursor & TO15 analysis

- Canister or on-line sampling of 108 pollutant compounds
- Cryogen free pre-concentration AND cryogen free chromatography
- 100% RH - VVOCs, OVOCs and VOCs in a single analysis
- Sub ppb detection limits from a 400mL sample



Ethene, Acetylene, Ethane to FID

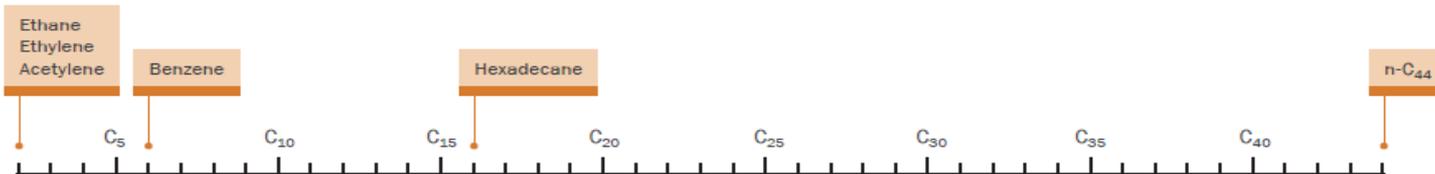
C<sub>3</sub> onwards to Mass Spec

# Challenges with air sampling

1. Many compounds are harmful at very low concentrations
  - ✓ Pre-concentration
2. Many pollutants are very volatile – how can we collect and concentrate them?
  - ✓ Cryogen – free on-line or canister sampling
  - ✓ Selection of the optimum sorbent configurations
3. Air samples can be very humid – how do we remove water without removing compounds?
  - ✓ Water management technologies – sorbent selection, dry purge, membrane drying or kori-xr
4. I need to analyse SVOCs as well as VOCs
5. I need to sample at numerous remote locations



# Techniques for air sampling and analysis



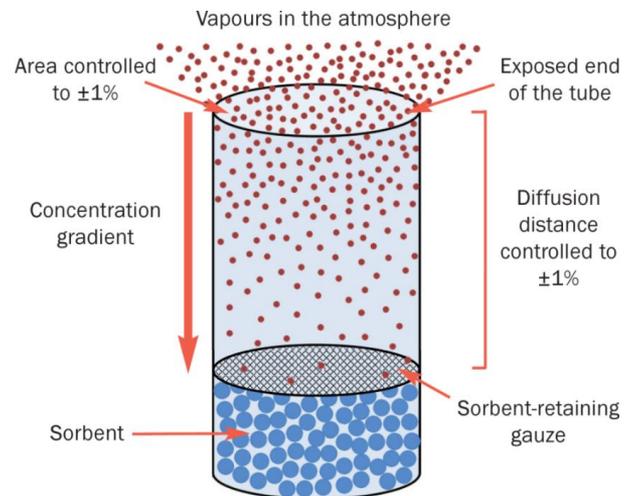
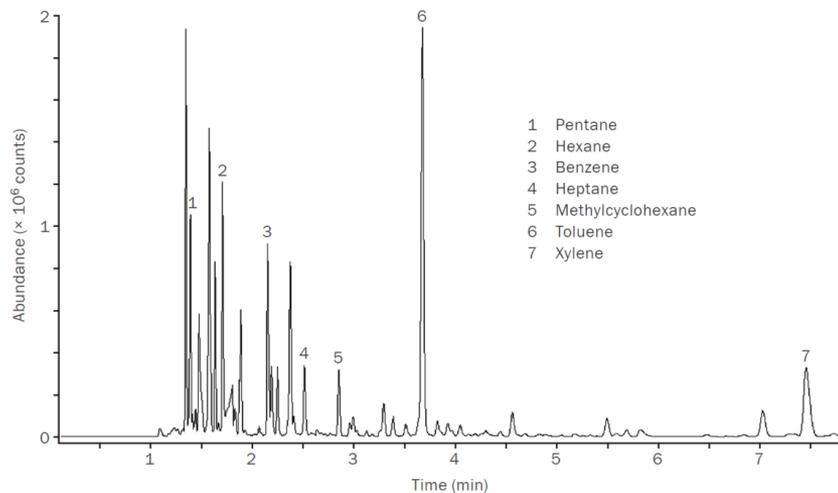
Canisters

On-line

Sorbent Tubes

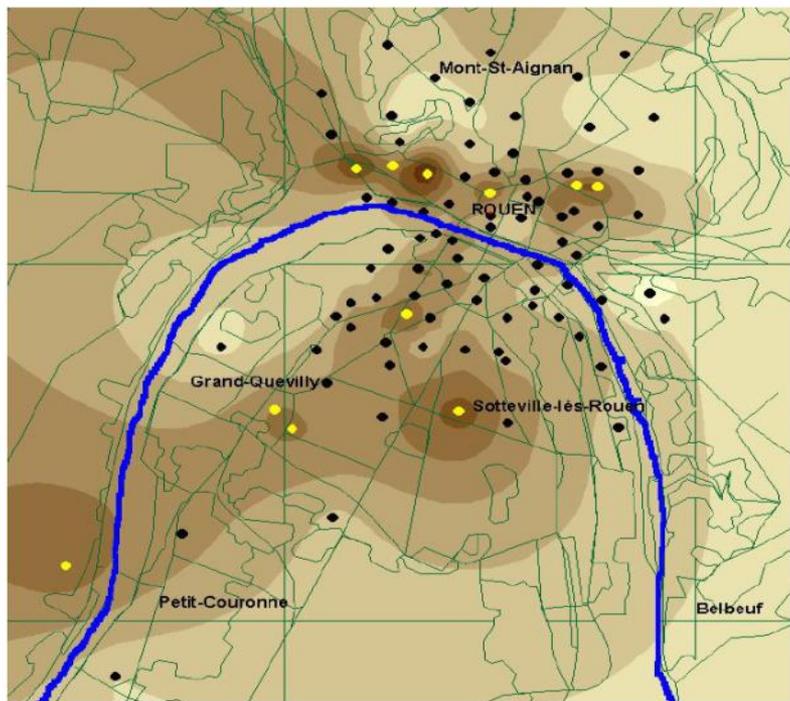


# Sorbent tube sampling: Passive



# Sorbent tube sampling: Passive

Five-day monitoring of benzene near Rouen, France



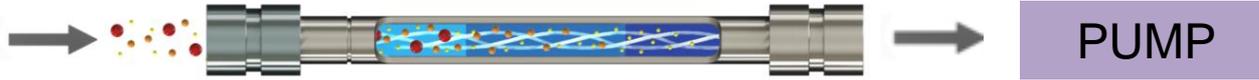
9 8 7 6 5 4 3  
Benzene concentration ( $\mu\text{g}/\text{m}^3$ )

- 5 days
- Over 90 passive samplers
- Easy to deploy
- Unattended sampling
- 100 tube autosampler = unattended analysis



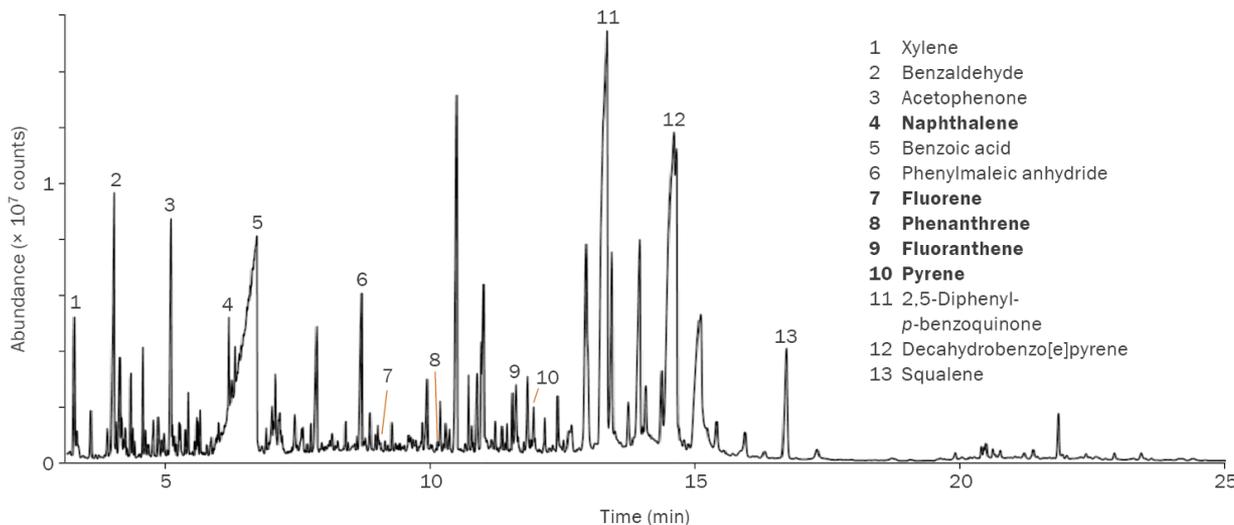
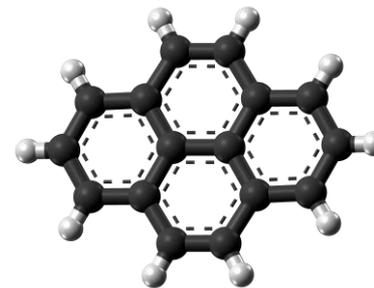
# Sorbent tube sampling: Pumped

- A pump actively draws the air through the sorbent tube



# Sorbent tube sampling: Pumped

- **180L** of Shanghai air collected on a sorbent tube
- 250 ml/min sampling for 12 h
- Many PAH compounds identified



# Challenges with air sampling

1. Many compounds are harmful at very low concentrations
  - ✓ Pre-concentration
2. Many pollutants are very volatile – how can we collect and concentrate them?
  - ✓ Cryogen – free on-line or canister sampling ( $C_2 - C_{14}$ )
  - ✓ Selection of the optimum sorbent configurations ( $C_2 - C_{44}$ )
3. Air samples can be very humid – how do we remove water without removing compounds?
  - ✓ Water management technologies – dry purge, membrane drying or kori-xr
4. I need to analyse SVOCs as well as VOCs
  - ✓ Sorbent tube sampling ( $C_{2/3} - C_{44}$ )
5. I need to sample at many remote locations
  - ✓ Sorbent tube sampling ( $C_{2/3} - C_{44}$ )

# The complete air monitoring solution

- ✓ Cryogen-free trapping of very volatile C2 hydrocarbons
- ✓ Automated analysis of 100 sorbent tubes AND up to 27 canisters or on-line samples in the same sequence
- ✓ Advanced water removal with dry purge and kori-xr



# Wider application of Thermal desorption



# Consumer environmental health (CEH)

What is the focus?

- Potential health risks that arise from poor indoor air quality have prompted regulation of a wide range of materials
- The average American spends 93% of their time indoors...
- 86% in houses, offices, etc...
- ... and 7% in automobiles.
- The majority of items which can effect the air quality in these locations are now regulated



# Sampling and Analysis options

## 1 Indoor air monitoring



**A** Pumped sampling

## 2 Final product certification



**B** Small chambers

## 3 Screening of emissions from components

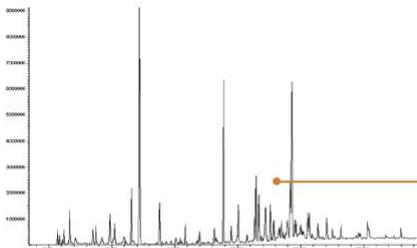


**C** Microchambers



**D** Direct desorption

The four main sampling approaches for assessing emissions from products and materials are all compatible with TD.



Analysis by gas chromatography

### TD-based approaches can detect:

- VOCs (boiling up to  $n\text{-C}_{10}$ ).
- SVOCs or FOG compounds (boiling above  $n\text{-C}_{10}$ ).
- Odorous or reactive species.

### Formaldehyde monitoring

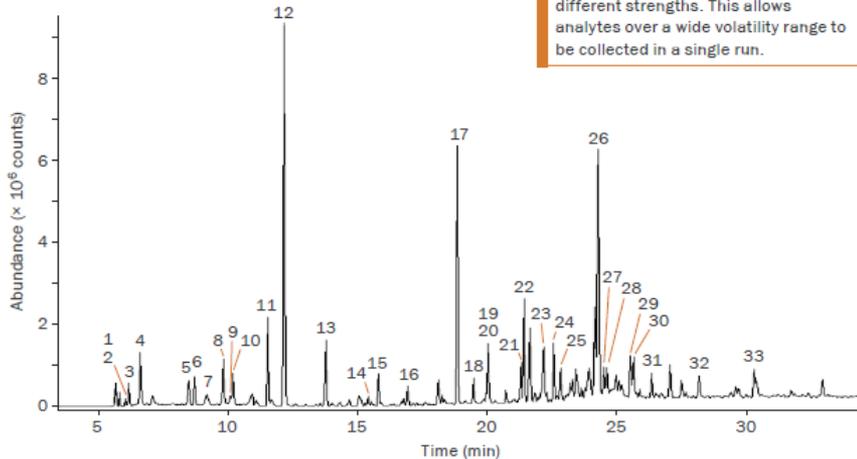
Note that formaldehyde and other aldehydes, unlike other VOCs/SVOCs, are analysed by HPLC rather than GC-MS, and require special sampling and analysis equipment such as DNPH cartridges. See page 20 for more information.



# Indoor air quality

## Assessing the cause of poor indoor air quality

1 1,1-Difluoroethane	14 Tetrachloromethane	27 Limonene
2 1,1,1,2-Tetrafluoroethane	15 Benzene	28 Undecane
3 Dichlorodifluoroethane	16 Methylcyclohexane	29 Benzyl alcohol
4 1,1-Difluoro-1-chloroethane	17 Toluene	30 Nonanal
5 Isopentane	18 Hexanal	31 Dodecane
6 Ethanol	19 Tetrachloroethane	32 Tridecane
7 Trichlorofluoromethane	20 Ethylcyclohexane	33 Tetradecane
8 Propanol	21 Ethylbenzene	
9 Freon® 113	22 <i>o</i> - <i>p</i> -Xylene	
10 Acetone	23 Styrene	
11 Dichloromethane	24 $\alpha$ -Pinene	
12 Hexane	25 Decane	
13 Ethyl acetate	26 Trimethylbenzene	

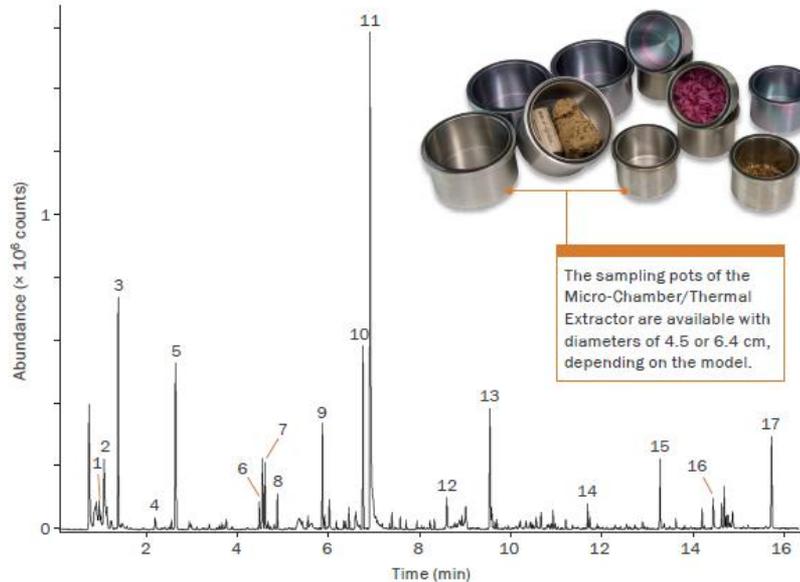


- Indoor pollutants typically arise from sources such as:
  - construction materials,
  - furnishings,
  - cleaning products
  - and consumer goods,
  - As well as activities such as cooking and smoking

**A wide range of potentially harmful chemicals** at ppt and ppb levels are identified in this indoor air sample, collected by pumped sampling onto sorbent tubes over relatively short time periods.

# Indoor air quality

## Assessing the cause of poor indoor air quality



1 Butane	8 Octanal	14 n-Hexadecanoic acid
2 Acetic acid	9 Nonanal	15 2,6-Diphenylphenol
3 Benzene	10 Decanal	16 Diisooctyl phthalate
4 Toluene	11 2-Phenoxyethanol	17 Squalene
5 Hexamethylcyclotrisiloxane	12 Geranyl acetone	
6 Benzaldehyde	13 2,2,4-Trimethyl-1,3-pentanediol diisobutanoate	
7 Phenol		

- To more accurately assess the source materials can be put into chambers to rate the emissions.
- As part of EU regulations products used in the constructions of homes must VOCs below a set threshold to ensure indoor air concentrations do not become harmful to those residing in them.

**Reflecting the versatility of TD**, a range of VOCs and SVOCs – from butane ( $C_4H_{10}$ ) to squalene ( $C_{30}H_{50}$ ) – are identified in this headspace-TD analysis of mahogany using the Micro-Chamber/Thermal Extractor.

# Breath sampling

How it works

- Alveolar ('end-tidal') breath collected and transferred onto sorbent tube
  - Provides information on VOC levels in blood (Pollutants and metabolites)

## ➤ Markes' Bio-VOC



## ➤ ReCIVA breath sampler (Owlstone Medical)

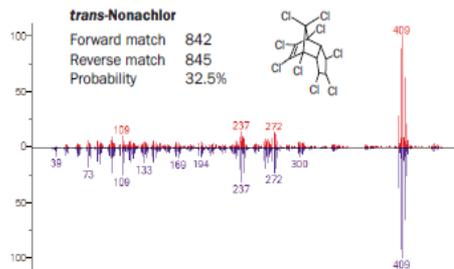
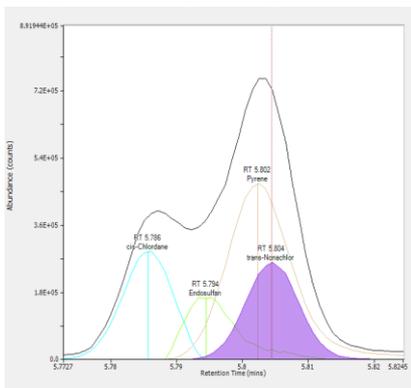
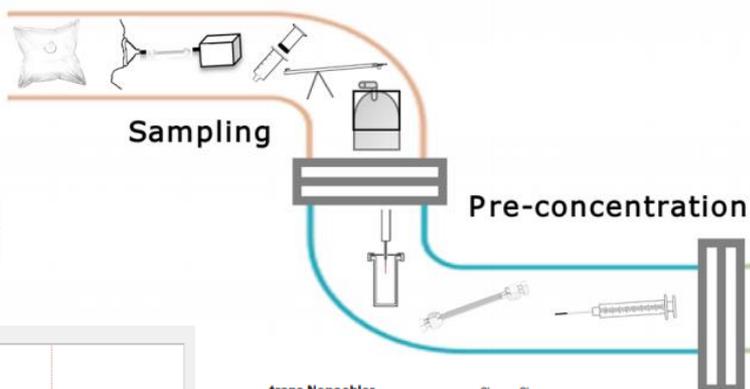


©

# Challenges with breath sampling

1. Compounds in breath are often at very low concentrations
  - ✓ Pre-concentration
2. Breath samples can be very humid – how do we remove water without removing compounds?
  - ✓ Water management technologies – sorbent selection & dry purge
3. I need to analyse SVOCs as well as VOCs
  - ✓ Sorbent tube sampling ( $C_{2/3} - C_{44}$ )
4. I need to sample at many remote locations
  - ✓ Sorbent tube sampling ( $C_{2/3} - C_{44}$ )

# Analytical challenges in breath analysis

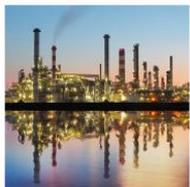


Measurement

Data Analysis

Source: Lawal et al., *Exhaled breath analysis: a review of 'breath-taking' methods for off-line analysis* (2017) *Metabolomics*, 13:110

# Application areas



Environmental monitoring

Biological profiling

Automotive studies

Defence & Forensics

Fragrance & odour profiling

Consumer environmental health

Food & Drink

## Contact Markes



enquiries@markes.com



+44 (0)1443 230935



www.markes.com



@MarkesInt



www.linkedin.com/company/markes-international

Since 1997

