VOC Purge & Trap

Determining Volatile Organic Compounds from Difficult Soil Matrices Utilizing Both Headspace and Purge and Trap Techniques

Abstract

Purge and Trap concentration (P&T) along with Gas Chromatography/Mass Spectrometry (GC/MS) analysis is the traditionally used method for the analysis of Volatile Organic Compounds (VOCs). This application note investigates P&T concentration of VOCs in soils utilizing both the standard soil and the automated methanol extraction technique of Teledyne Tekmar's Atomx, a combination P&T concentrator and multimatrix autosampler system. Furthermore, we will demonstrate an alternative method for the analysis of VOCs in soils utilizing both the Static and Dynamic modes of Teledyne Tekmar's HT3™ Automated Headspace Analyzer. The analysis of soils presents several challenges as the composition of the soil matrix can vary. Thus, this application note will also inquire into different soil matrices and how to best determine the VOCs present in them.

Introduction

The potential for soils to be contaminated with VOCs can be quite high especially if the soil sample is near a commercial area, a landfill, a manufacturing facility or a host of other potentially contaminated sites. The amount of contamination can vary from parts per billion (ppb) to pure product contamination. The type of soil being contaminated can also vary from sand, to clay, to a humic soil. In order to analyze VOCs in a soil matrix it becomes necessary to account for all of these variances.

In this study, sand, clay and humic soil were analyzed via P&T concentration, utilizing the Atomx, and by Headspace analysis, using the HT3™ Automated Headspace Analyzer. The results of these analyses were then compared in order to show the benefits and the possible problems with these analyses and their respective levels of detection. From these examinations, experimental and instrumental conditions were optimized and recommendations on the type of analysis that would best suit the level of contamination and the type of soil matrix being examined.

Definitions

1. **Static Headspace**

A soil sample is placed in a sealed vial and thermostatically controlled for a fixed amount of time. Once equilibration is achieved between the sample and gaseous phases a sample of the gas phase is introduced to the GC/MS for separation, identification, and quantification.

2. **Dynamic Headspace**

The same procedure as Static is employed with the exception being the vial is not permitted to reach equilibrium and instead the headspace is continually swept onto an adsorbent trap where the compounds of interest are concentrated. After a certain amount of time the adsorbent trap is heated and back flushed to the GC/MS.

3. **In Vial Purge**

A soil sample is placed in a sealed vial containing a predetermined amount of laboratory grade water and a magnetic mixing bar. An inert purge gas is introduced directly into the soil water slurry while the sample is mixing via a two stage sampling needle. The purge gas exits the vial extracting the compounds of interest and depositing them onto an adsorbent trap. Then as in Dynamic Headspace the analytical trap is heated and back flushed to the GC/MS.

4. **Solvent Extraction**

A soil sample is placed in a vial containing a predetermined amount of solvent and mixed for a fixed amount of time. Once the compounds have diffused from the soil and into the solvent, an aliquot of the extract is transferred into laboratory grade water. The water and extract are then purged similarly as to the In Vial Purge with the difference being they are placed in a fritted glass sparger that increases surface interaction via the frit which finely divides the incoming purge gas. Desorption and transfer to the GC/MS is the same as the previously mentioned three techniques.

Experimental-Instrument Conditions

The Atomx P&T Concentrator/Multimatrix Autosampler and the HT3™ Automated Headspace Analyzer were coupled to an Agilent 7890A GC and a 5975C inert XL MSD for these analyses. The Atomx and the HT3™ were both equipped with a #9 adsorbent trap. Tables 1 and 2 display the GC, MSD conditions while Tables 3 and 4 display the Atomx conditions for low and high level VOC analysis. The HT3™ Automated Headspace Analyzer Static and Dynamic experimental conditions are listed in Tables 5 and 6 respectively.

Tables 1 & 2: GC and MSD Parameters

Table 3: Atomx High Level, Solvent Extraction, VOC Parameters (Parameters highlighted in yellow were varied and dependant on the soil matrix.)

Table 4: Atomx Low Level VOC Parameters (Parameters highlighted in yellow were varied and dependant on the soil matrix.)

HT3 Static Parameters			
Variable	Value	Variable	Value
Constant Heat Time	On	Mixing Time	Varied
G.C. Cycle Time	30.00 min	Mixing Level	Varied
Valve Oven Temp	125° C	Mixer Stabilize Time	0.50 min
Transfer Line Temp	125° C	Pressurize	Varied
Standby Flow Rate	50mL/min	Pressurize Time	1.50 min
Platen/Sample Temp	Varied	Pressurize Equil. Time	0.50 min
Platen Temp Equil. Time	1.00 min	Loop Fill Pressure	Varied
Sample Equil. Time	Varied	Loop Fill Time	0.50 min
Mixer	On	Inject Time	1.00 min

Table 5: HT3™ Static Headspace Parameters (Parameters highlighted in yellow were varied and dependant on the soil matrix.)

Calibration

A 50ppb working calibration stock standard was prepared in methanol. The low level soil curve range for the in vial purge analysis utilizing Purge and Trap was 1.0-200ppb. Solvent extraction was used for the high level soil curve. The range for the solvent extraction curve was also 1.0-200ppb as the samples are at a 100 times dilution. The static headspace curve had a range of 50ppb-2.0ppm and the dynamic headspace curve range was 5.0ppb-200ppb. All of the calibration data was analyzed using Agilent Chemstation software.

Results

Sand Matrix

Baked sand was used for the sand matrix. There were a few advantages to the sand matrix the first being that the VOCs extracted into the solvent very well when using the P&T solvent extraction technique. Second most of the VOCs purged out of the sand during the in vial purge of the low level P&T analysis. Finally,

most of the VOCs swept into headspace well during static and dynamic headspace analysis however, de-ionized water needed to be added to the matrix in order to better release the compounds into the headspace. The main disadvantage of the sand matrix was the interaction of the polar VOCs with the sand. These interactions caused inconsistent recoveries of these compounds during both static and dynamic headspace analysis, see figures 1 and 2. In addition, compounds with higher boiling points showed less recoveries when compared with the aqueous standard for the in vial purge as seen in figure 3. Solvent extraction, however, exhibited no issues with any of the VOCs and gave consistent recoveries for all of the compounds analyzed, see figure 4.

Figure 1: Overlay of 1ppm Static Headspace Sand Matrix Sample (red) and Static Headspace VOC Cal Standard (blue)

Figure 2: Overlay of 100ppb Dynamic Headspace Sand Matrix (red) Sample and Dynamic Headspace VOC Cal Standard (blue)

Figure 3: 50ppb Chromatogram Overlay of Low VOC Sand (red) and Low VOC Cal Standard (blue)

Figure 4: 50ppb Chromatogram Overlay of Solvent Extracted VOC Sand (red) and High VOC Cal Standard (blue)

Clay Matrix

Natural clay was used for the clay matrix. The clay matrix proved to have several advantages. The VOCs went into headspace very well with the addition of de-ionized water. However, due to some breakthrough, there was some loss of compound recovery with the lighter compounds, see figures 5 and 6. In addition, the VOCs extracted well for both the low level, in vial purge, and high level, solvent extraction, P&T analyses with no loss of the lighter VOCs, see figures 7 and 8. The central drawback for extracting VOCs out of a clay matrix for both headspace and P&T analyses was the longer agitation times and more vigorous mixing required for compound extraction.

Figure 5: Overlay of 1ppm Static Headspace Clay Matrix Sample (red) and Static Headspace VOC Cal Standard (blue)

Figure 6: Overlay of 100ppb Dynamic Headspace Clay Matrix (red) Sample and Dynamic Headspace VOC Cal Standard (blue)

Figure 7: 50ppb Chromatogram Overlay of Low VOC Clay (red) and Low VOC Cal Standard (blue)

Figure 8: 50ppb Chromatogram Overlay of Solvent Extracted VOC Clay (red) and High VOC Cal Standard (blue)

Soil Matrix

Humic soil was used for the soil matrix. The humic soil was very absorbent and proved to be the most difficult matrix to extract VOCs from. The only advantage to this matrix is that the VOCs extracted into solvent very well and proved to have consistent recoveries, see Figure 12. However, there were several disadvantages to this matrix. First of all, the soil readily absorbed both de-ionized water and solvent, so all of the samples needed to be diluted further than the sand and clay matrices. Secondly, since the soil was light and did not settle well after mixing, the settle times and the sweep times had to be increased in order to achieve better VOC recoveries. In the case of P&T, a longer time was required to sweep the needle along with a larger solvent rinse of the needle in order to ensure system cleanliness. Finally, the matrix interfered with the heavier VOCs migration into headspace, see figures 9 and 10, and the purge efficiency of the heavier compounds for low level VOC P&T, see figure 11. Thus, the recoveries for the heavier VOCs were inconsistent for all of the analyses with the exception being solvent extraction.

Figure 9: Overlay of 1ppm Static Headspace Soil Matrix (red) and Static Headspace VOC Cal Standard (blue)

Figure 10: Overlay of 100ppb Dynamic Headspace Soil Matrix (red) Sample and Dynamic Headspace VOC Cal Standard (blue)

Figure 11: 50ppb Chromatogram Overlay of Low VOC Soil (red) and Low VOC Cal Standard (blue)

Figure 12: 50ppb Chromatogram Overlay of Solvent Extracted VOC Soil (red) and High VOC Cal Standard (blue)

Conclusions

The HT3™ Automated Headspace Analyzer offered several advantages in measuring VOCs in a soil matrix. The system can measure both high amounts of contamination utilizing the Static Headspace option and low amounts of contamination using the Dynamic Headspace option. The system also has the advantage of measuring high amounts of VOCs utilizing the Static Headspace option without the carryover or contamination problems often seen with purge and trap analysis. The HT3™ Automated Headspace Analyzer can run both static and dynamic headspace analysis within the same schedule and the system has the Method Optimization Mode (M.O.M.) which can automatically modify headspace parameters in order to determine optimum experimental conditions. The different soil matrices however did pose some compound recovery issues when using the headspace analyzer and matrix would have to be a consideration when doing this analysis. The Atomx P&T concentrator/Multimatrix Autosampler is an excellent system for determining VOCs in a soil matrix. This system has the distinct advantage of being able to run low level, in vial purge, P&T and also has an automated solvent extraction option for P&T. For low level VOC analysis, the in vial purge technique detected VOCs very well in clay, however both the sand and soil matrices affected compound recoveries. Solvent extraction on the other hand, proved to be the best choice in determining VOCs in difficult matrices as there were no issues with compound recovery no matter what matrix was being analyzed.

